

January 2016

January 2016						
◀ December 2016						February 2016 ▶
Sun	Mon	Tue	Wed	Thu	Fri	Sat
					1	2
3	4 Day 6 Density Solids vs. liquids CALCULATIONS	5 Day 1 Density of a solid and a liquid pre-lab Practice Calculations	6 Day 2 Density of a solid and a liquid Lab Analyze and write up conclusion	7 Day 3 Review for Test	8 Day 4 Test -Classification of particles (atoms vs molecules; elements vs. compounds) - Particle model -Pure substances vs. mixtures - Homo vs. Hetero mixtures - Physical Properties of Matter – melting point, boiling point, solubility and density. -Effects of Temperature on Solubility - Density – how to calculate for a solid and a liquid	9
10 Week 1	11 Day 5 Review for Lab Exam	12 Day 6 Lab Exam Day 1	13 Day 1 Lab Exam Day 2	14 Day 2 Review for Midyear Exam	15 Day 3 Review for Midyear Exam	16
17 Week 2	18 Day 4 EXAM WEEK	19 Day 5 Gr.9 Exam Chemistry Exam	20 Day 6 EXAM WEEK	21 Day 1 EXAM WEEK	22 PED DAY	23

Sun	Mon	Tue	Wed	Thu	Fri	Sat
<p>24</p> <p>Week 3</p>	<p>25 Day 2</p> <p>Characteristic Chemical Properties pH scale</p> <p>Indicators Gas Tests Demo – H₂, O₂, CO₂</p> <p>Indicators Pre- Lab</p>	<p>26 Day 3</p> <p>Indicators Lab</p>	<p>27 Day 4</p> <p>Complete Indicators Lab</p> <p>Physical vs. Chemical Changes of Matter</p> <p>Physical Changes -review of all the phase changes & connect to particle model</p> <p>Co- teach</p>	<p>28 Day 5</p> <p>What is a chemical reaction?</p> <p>Evidence for a chemical change</p> <p>Co- teach</p>	<p>29 Day 6</p> <p>END of Term 2 Go over the exams</p> <p>Clean out folders... term 3 graph</p>	<p>30</p>
<p>31</p>	<p>Notes: Electrolysis Demo</p>					